

CATALYTIC PROCESS FOR EFFICIENT ENANTIODIVERGENCE OF MESO-N,N'-DIACETYL-2-IMIDAZOLIDINONES AND DL-N-ACETYL-2-OXAZOLIDINONES

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Abstract: A reductive monodeacylation, catalyzed by oxazaborolidines derived from conformationally rigid chiral aminoalcohols provideds a practical method for the effective enantiodivergence of meso-1,3-diacetyl-2-imidazolidinones. This catalytic deacylation is successfully applied to the kinetic resolution of racemic 1-acetyl-2-oxazolidinones. © 1998 Elsevier Science Ltd. All rights reserved.

Sterically congested and conformationally fixed 2-oxazolidinones¹ and 2-imidazolidinones² such as the tricyclic compounds 1-5 have proven to be excellent chiral auxiliaries for use in highly enantiocontrolled carbon-carbon bond formation. The conventional method for the preparation of such types of chiral heterocycles involves an optical resolution step using (1S,2R)-2-methoxy-1-apocamphanecarboxylic acid (MAC-acid),³ which is efficient, but tedious and time-consuming. An alternative procedure which has recently been reported for the enantiodivergence of 1,3-diacetyl-2-imidazolidinones also requires stoichiometric amounts of the chiral reagent 7 as the lithium salts.² As a result, there is an ongoing need for a practical method for the large scale preparation of such types of chiral 2-imidazolidinone and 2-oxazolidinone auxiliaries via a catalytic process.



1 (X=O, DMAOx)
2 (X=N-R, DMAIm)

NH X O

3 (X=O, DHAOx)
4 (X=N-R, DHAIm)

NH O

5 (X=O, HMCOx)6 (X=N-R, HMCIm)

We wish to report herein a catalytic procedure for the efficient enantiodivergence of *meso-1*, 3-diacetyl-2-imidazolidinones and *dl-1*-acetyl-2-oxazolidinones by a borane-mediated reductive monodeacetylation, which is catalyzed by the sterically constrained 2-aminoalcohols, 7 and 8, which are, in turn, readily prepared by the ring-opening of the chiral 2-oxazolidinone auxiliaries, 3 and 5, respectively.



NH₂ OH



0040-4039/98/\$ - see front matter © 1998 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(98)01340-9 We recently reported that the conformationally ridid tricyclic oxazaborolidines, derived from *cis*-fixed (+)-aminoalcohols 7 and 8, have great potential as efficient catalysts for the enantioselective borane-reduction of ketones.⁴ This borane-mediated reduction has been successfully applied to the enantioselective monodeacylation of *meso*-1,3-diacyl-2-imidazolidinones, derived from the cycloaddition of 1,3-diacyl-2-imidazolones to anthracenes, and which can be readily monodeacylated.²

Thus, the borane-mediated monodeacetylation of meso-1,3-diacetyl-2-imidazolidinone 10a (R=H) proceeded smoothly at room temperature to give (+)-1-acetyl-2-imidazolidinone $11a^5$ with excellent enantioselectivity, in excess of 99% ee, when (+)-aminoalcohol 7 (5-10 mol%) was used as a catalyst, in combination with borane-methyl sulfide complexes. In a similar manner, the more bulky meso-compound 10b (R=Me) also underwent smooth monodeacetylation to give (+)-1-acetyl-2-imidazolidinone $11b^6$ with 98% ee. The stereochemistry of the 11a and b, which are preferentially formed has been previously established. The sterically more congested (+)-aminoalcohol 8 was moderately effective, as seen in Table 1, while the B-methyl oxazaborolidine derived from $(S)-\alpha$, α -diphenyl-2-pyrrolidinemethanol (9) and trimethylboroxine was much less effective as a catalyst. Thus, the aminoalcohol 7 appears to be the chiral reagent of choice for

Table 1 Enantioselective Monodeacetylation of *Meso-1*, 3-Diacetyl-2-imidazolidiones (10a, b) Catalyzed by Oxazaborolidines Derived from Aminoalcohols (7, 8)

R (10)	Aminoalcohol (mol %)	Borane (equiv.)	Time (h)	Yield (%)	% ee b)
Н	7 (10)	BH ₃ (0.7)	2	74	99 (11a)
	7 (5)	BH ₃ (0.65)	2	69	99 (11a)
	8 (10)	BH ₃ (0.7)	4	31	90 (11a)
	9 (10) ^{c)}	BH ₃ (0.6)	4	9	78 (12a)
	7 (10)	BH ₃ (0.15) +thexylborane (2)	8	78	99 (11a)
Me	7 (10)	BH ₃ (0.7)	2	71	98 (11b)
	7 (10)	BH ₃ (0.15) +thexylborane (2)	8	81	98 (11b)

a) Isolated yield. b) Determined by HPLC analysis. c) With the B-methyl oxazaborolidine derived from 9 and trimethylboroxine.

the enantiodivergence of meso-10.

This reductive deacetylation was accompained, as expected, by the formation of by-products such as N-ethyl- and N, N'-diethyl-2-imidazolidinones in 5-10% yield, depending on the reaction conditions. When the more bulky reducing agent, thexylborane, was used in place of BH₃-complexes, the side reactions were greatly suppressed, resulting in negligible amounts of the N-ethyl compounds. Under the modified conditions and employing thexylborane in the presence of catalytic amounts of BH₃-complexes, the yield was appreciably improved with no decrease in enantioselectivity (Table 1).⁸ The optically active 1-acetyl-2-imidazolidinones 11a and b thus obtained, after purification by a single recrystallization, serve as excellent precursors for chiral 2-imidazolidinone auxiliaries as has been recently demonstrated.²

This reducing system was sufficiently effective to permit the facile monodeacetylation of diacetyl-imidazolidinone 13 without affecting olefinic function, thus providing a good precursor 14⁹ for the chiral auxiliaries 6 with excellent enantioselectivity, in excess of 98% cc (Scheme 2).

Interestingly, this catalytic deacylation process was versatile enough to permit the kinetic optical resolution of the racemic 3-acetyl-2-oxazolidinones. Thus, the cycloadduct 15b (R=Me) derived from 3-acetyl-2-oxazolone and 9, 10-dimethylanthracene was treated with 0.9 equimolar amounts of borane-methyl sulfide complexes in the presence of the chiral aminoalcohol 7 (10 mol%) at room temperature for 4 h to give (+)-N-acetyl-2-oxazolidinone 16b¹⁰ in 99% ee in 41% yield, in addition to the deacetylated derivative 17b in 72% ee in 56% yield. The use of less amounts of the borane complex (0.6 equiv.) considerably enhanced the enantioselectivity of 17b, up to 90% ee, but in a reduced yield (below 20%). Exceptionally large differences in the rate between the enantiomers clearly demonstrate the potential of this method for nonenzymatic kinetic resolution. In a similar manner, the anthracene-derived cycloadduct 15a (R=H) gave 16a¹¹ in 98% ee, as seen in Scheme 3.

In both cases, the formation of 3-ethyl-2-oxazolidinones, as might be expected, were not detected.

Deacylation of the 16a and b thus obtained with cesium carbonate followed by single recrystallization

gave, respectively, the pure 2-oxazolidinone auxiliaries, (+)-DHAOx ^{1a} and (+)-DMAOx. ^{1b}

In conclusion, the catalytic monodeacylation process presented here is highly practical for the preparation of versatile chiral tricyclic auxiliaries, and may be applicable to the facile synthesis of chiral building blocks for vis-diarnine and vic-aminoalcohol skeletons which are found in a substantial number of bioactive compounds.

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- 5. **11a** (R=H): mp 245.0 °C (from EtOH), $[\alpha]_D + 123.3$ ° (c 1.0, CHCl₃).
- 6. **11b** (R=Me): mp 235.0 °C (from MeOH), $[\alpha]_0 + 165.2$ ° (c 1.0, CHCl₃).
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- 8. Typical procedure for enantioselective monodeacylation:
 - a) To a stirred solution of 7 (0.05 mmol) and $BH_3 \cdot SMe_2$ (0.1 mmol) in THF (2 ml) was added a solution of 10a (0.5 eq) and $BH_3 \cdot SMe_2$ (0.25 mmol) in THF (4 ml) at 0 °C under an atmosphere of argon. After stirring for 2h at room temperature, the mixture was acidified with 3N HCl. The usual work-up, followed by chromatographic purification, gave (+)-1-acetyl-2-imidazolidinone 11a, in 74% yield whose optical purity was determined to be 99% ee by HPLC analysis on a Chiralcel OD-H column.
 - b) A solution of 7 (0.05 mmol) and $BH_3 \cdot SMe_2$ (0.1 mmol) in THF (2 ml) was stirred at room temperature for 15 min. and 10b (0.5 mmol) was then added at 0 °C. To the mixture, a 0.4 M solution of thexylborane (1 mmol) in THF was added dropwise, followed by stirring at room temperature for 8 h. The work-up, as above, gave an 81% yield of 11b in 98% ee.
- 9. **14**: mp 154.0 °C (from hexane), $[\alpha]_D + 161.2^\circ$ (c 1.0, CHCl₃), ¹H-NMR (500 MHz / CDCl₃) δ : 5.86 (1H, brs), 4.63 (1H, d, J=8.5 Hz), 3.75 (1H, d, J=8.5 Hz), 2.42 (3H, s), 1.58 (3H, s), 1.49 (3H, s), 1.03 (3H, s), 0.73 (3H, s), 0.63 (3H, s). The absolute configuration is tentatively assigned, based on comparison with **11a** and **b**.
- 10. **16b** (R=Me): mp 213 °C (from hexane-CH₂Cl₂), $[\alpha]_D$ +204.8° (c 1.0, CHCl₃), ¹H-NMR (270 MHz / CDCl₃) δ : 7.32 (8H, m), 4.62 (1H, d, J=8.4 Hz), 4.49 (1H, d, J=8.4 Hz), 2.41 (3H, s), 2.09 (3H, s), 1.93 (3H, s).
- 11. **16a** (R=H): mp 171 °C (from hexane-CH₂Cl₂), $[\alpha]_D$ +163.6° (c 1.0, CHCl₃), ¹H-NMR (270 MHz / CDCl₃) δ : 7.32 (8H, m), 5.05 (1H, d, J=3.5 Hz), 4.78 (1H, dd, J=3.5, 8.5 Hz), 4.68 (1H, d, J=3.5 Hz), 4.55 (1H, dd, J=3.5, 8.5 Hz), 2.32 (3H, s).